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Chapter 12 Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. ____ 1. The first step in most stoichiometry problems is to _____. a. add the coefficients of the reagents c. convert given quantities to volumes b. convert given quantities to moles d. convert given quantities to masses ____ 2.

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Chapter 12 Stoichiometry 127 SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353-358) This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

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Part 12.11A : First, we will calculate the theoretical yield based on the stoichiometry. Step 1: List the known quantities and plan the problem. Known . given: mass of $\text{KClO}_3 = 40.0 \text{ g}$; molar mass $\text{KClO}_3 = 122.55 \text{ g/mol}$; molar mass $\text{O}_2 = 32.00 \text{ g/mol}$; Unknown . theoretical yield $\text{O}_2 = ? \text{ g}$

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