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1. What is the number of atoms of oxygen in 2.0 mol of hydrated sodium carbonate,  $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$ ? Avogadro's constant,  $L$  or  $N_A$  is  $6.02 \times 10^{23} \text{ mol}^{-1}$ .  
A. 6 B. 26 C.  $103.6 \times 10^{23}$  D.  $101.6 \times 10^{23}$
2. What is the volume, in  $\text{cm}^3$ , of the final solution if  $100 \text{ cm}^3$  of a solution containing 1.42 g of sodium sulfate,  $\text{Na}_2\text{SO}_4$ , is diluted to the ...

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- The volume occupied by one mole of an ideal gas at 273 K and  $1.01 \times 10^5 \text{ Pa}$  is  $22.4 \text{ dm}^3$ . What volume, in  $\text{dm}^3$ , is occupied by 3.20 g  $\text{O}_2$  (g) at 273 K and  $1.01 \times 10^5 \text{ Pa}$ ?  
A. 2.24 B. 4.48 C. 22.4 D. 71.7

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1 H 1.01 3 Li 6.94 11 Na 22.99 19 K 39.10 37 Rb 85.47 55 Cs 132.91 87 Fr (223) Atomic Number Element Atomic Mass 25 Mn

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